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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/669,713	09/25/2003	Satoru Fukuoka	031212	6383
38834	7590 10/27/2006	•	EXAMINER	
WESTERMAN, HATTORI, DANIELS & ADRIAN, LLP			ECHELMEYER, ALIX ELIZABETH	
SUITE 700	ECTICUT AVENUE, NW		ART UNIT	PAPER NUMBER
WASHINGTON, DC 20036			1745	
			DATE MAILED: 10/27/2006	5

Please find below and/or attached an Office communication concerning this application or proceeding.

			<i></i>
	Application No.	Applicant(s)	
	10/669,713	FUKUOKA ET AL.	
Office Action Summary	Examiner	Art Unit	
	Alix Elizabeth Echelmeyer	1745	
The MAILING DATE of this communication app Period for Reply	pears on the cover sheet with the c	correspondence address	
A SHORTENED STATUTORY PERIOD FOR REPL WHICHEVER IS LONGER, FROM THE MAILING D. - Extensions of time may be available under the provisions of 37 CFR 1.1 after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period realiure to reply within the set or extended period for reply will, by statute Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION (36(a). In no event, however, may a reply be tirwill apply and will expire SIX (6) MONTHS from the course the application to become ABANDONE	N. nely filed the mailing date of this communication (35 U.S.C. § 133).	
Status			
1)⊠ Responsive to communication(s) filed on <u>06 C</u>	October 2006.		
,_	s action is non-final.		
3) Since this application is in condition for allowa			is
closed in accordance with the practice under be	Ex parte Quayle, 1935 C.D. 11, 4	53 O.G. 213.	
Disposition of Claims			
4)⊠ Claim(s) <u>1-5</u> is/are pending in the application.			
4a) Of the above claim(s) is/are withdra	wn from consideration.		
5) Claim(s) is/are allowed.			
6)⊠ Claim(s) <u>1-5</u> is/are rejected.			
7) Claim(s) is/are objected to.			
8) Claim(s) are subject to restriction and/c	or election requirement.		
Application Papers			
9) The specification is objected to by the Examine	er.		
10) ☐ The drawing(s) filed on is/are: a) ☐ acc	cepted or b) objected to by the	Examiner.	
Applicant may not request that any objection to the	drawing(s) be held in abeyance. Se	e 37 CFR 1.85(a).	
Replacement drawing sheet(s) including the correct			
11) The oath or declaration is objected to by the Ex	xaminer. Note the attached Office	Action or form PTO-152.	
Priority under 35 U.S.C. § 119			
12) ☐ Acknowledgment is made of a claim for foreign a) ☐ All b) ☐ Some * c) ☐ None of: 1. ☐ Certified copies of the priority document	ts have been received.		
2. Certified copies of the priority document3. Copies of the certified copies of the priority			
application from the International Burea		ea in this Hational Otage	
* See the attached detailed Office action for a list		ed.	
	·		
Attachment(s)			
1) Notice of References Cited (PTO-892)	4) Interview Summary Paper No(s)/Mail D		
 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date 10-6-06. 	5) Notice of Informal F 6) Other:		

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DETAILED ACTION

Response to Amendment

1. This Office Action is in response to Applicants' After Final amendments filed October 6, 2006. Claim 1 has been amended. The arguments concerning the separator are found to be persuasive and will be entered. Claims 1-5 are pending and are rejected non-finally for the reasons given below.

Claim Rejections - 35 USC § 103

- 2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 3. Claims 1 and 3-5 are rejected under 35 U.S.C. 103(a) as being unpatentable over Hamrock et al. (US Patent 6,063,522) in view of Sano et al. (US Pre-Grant Publication 2002/0086191).

Regarding claim 1, Hamrock et al. teach a non-aqueous electrolytic solution for a lithium cell containing linear ethers such as diethylene glycol dimethyl ether (DMG) (column 13 lines 52-59).

As for applicants' claim 3, Hamrock et al. teach the use of conductive salts in the electrolyte composition (column 11 lines 45-50). Hamrock et al. list lithium bis (trifluoromethanesulfonyl) imide and lithium bis (pentafluoroethanesulfonyl) imide as preferred conductive salts (column 13 lines 20-25).

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Regarding claims 4 and 5, Hamrock et al. teach $Li_xMn_2O_4$ and Li_xMnO_2 as suitable cathode materials (column 14 lines 49-51).

Regarding claim 1, Hamrock et al. fail to teach the use of a separator having a melting point greater than 185 degrees Celsius.

Sano et al. teach the use of a separator in a battery cell that is capable of withstanding high temperatures ([0015]). Sano et al. teach that polyphenylene sulfide may be used as the separator, the same material used as the separator in the specification of the instant invention (claim 4 of Sano et al.).

Sano et al. further teach that the separator should be capable of withstand high temperatures in order to suppress the vaporization of the electrolyte ([0015]).

It would be desirable to use the polyphenylene sulfide separator of Sano et al. in the battery of Hamrock et al. in order to suppress the vaporization of the electrolyte.

Therefore, it would have been obvious to one having ordinary skill in the art at the time the invention was made to use the polyphenylene sulfide separator of Sano et al. in the battery of Hamrock et al. in order to suppress the vaporization of the electrolyte.

4. Claim 2 is rejected under 35 U.S.C. 103(a) as being unpatentable over Hamrock et al. in view of Sano et al. as applied to claim 1 above, and further in view of Takahashi et al. (JP 2003-249263).

The teachings of Hamrock et al. and Sano et al. as discussed above are incorporated herein.

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Hamrock et al. teach that mixtures of matrix materials can be tailored to provide optimum performance (column 14 lines 10-12). Hamrock et al. also teach to use of esters such as propylene carbonate and ethylene carbonate in the non-aqueous solvent (column 13 lines 52-63).

Hamrock et al. fail to teach the blended solvent including the cyclic ester carbonate as a subsidiary component.

Takahashi et al. teach that the solvent of the electrolyte of a lithium secondary battery contains propylene carbonate and diethylene glycol dialkyl ether (abstract). Takahashi et al. teach that the solvent may be 3-50% by volume propylene carbonate and 97-50% diethylene glycol dialkyl ether ([0014]).

Takahashi et al. further teach that this blend is desirable since it increases the conductivity of the electrolyte.

It would be desirable to use the blended solvent of Takahashi et al. in the battery of Hamrock et al. in view of Sano et al. in the amounts taught by Takahashi et al. in order to increase the conductivity of the electrolyte.

Therefore, it would have been obvious to one having ordinary skill in the art at the time the invention was made to use the blended solvent of Takahashi et al. in the battery of Hamrock et al. in view of Sano et al. in the amounts taught by Takahashi et al. in order to increase the conductivity of the electrolyte.

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Double Patenting

5. The nonstatutory double patenting rejection is based on a judicially created doctrine grounded in public policy (a policy reflected in the statute) so as to prevent the unjustified or improper timewise extension of the "right to exclude" granted by a patent and to prevent possible harassment by multiple assignees. A nonstatutory obviousness-type double patenting rejection is appropriate where the conflicting claims are not identical, but at least one examined application claim is not patentably distinct from the reference claim(s) because the examined application claim is either anticipated by, or would have been obvious over, the reference claim(s). See, e.g., *In re Berg*, 140 F.3d 1428, 46 USPQ2d 1226 (Fed. Cir. 1998); *In re Goodman*, 11 F.3d 1046, 29 USPQ2d 2010 (Fed. Cir. 1993); *In re Longi*, 759 F.2d 887, 225 USPQ 645 (Fed. Cir. 1985); *In re Van Omum*, 686 F.2d 937, 214 USPQ 761 (CCPA 1982); *In re Vogel*, 422 F.2d 438, 164 USPQ 619 (CCPA 1970); and *In re Thorington*, 418 F.2d 528, 163 USPQ 644 (CCPA 1969).

A timely filed terminal disclaimer in compliance with 37 CFR 1.321(c) or 1.321(d) may be used to overcome an actual or provisional rejection based on a nonstatutory double patenting ground provided the conflicting application or patent either is shown to be commonly owned with this application, or claims an invention made as a result of activities undertaken within the scope of a joint research agreement.

Effective January 1, 1994, a registered attorney or agent of record may sign a terminal disclaimer. A terminal disclaimer signed by the assignee must fully comply with 37 CFR 3.73(b).

6. Claims 1-3 are provisionally rejected on the ground of nonstatutory obviousness-type double patenting as being unpatentable over claims 1 and 4-6 of copending Application No. 10/787,749 in view of Sano et al.

Claims 1 and 4-6 of 10/787,749 teach the blended electrolyte of the instant invention but fail to teach a separator having a specific melting point.

Sano et al. teach the use of a separator in a battery cell that is capable of withstanding high temperatures ([0015]). Sano et al. teach that polyphenylene sulfide may be used as the separator, the same material used as the separator in the specification of the instant invention (claim 4 of Sano et al.).

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The separator of Sano et al. is desirable since it is capable of withstand high temperatures in order to suppress the vaporization of the electrolyte ([0015]).

It would be desirable to use the polyphenylene sulfide separator of Sano et al. in the battery of 10/787,749 in order to suppress the vaporization of the electrolyte.

Therefore, it would have been obvious to one having ordinary skill in the art at the time the invention was made to use the polyphenylene sulfide separator of Sano et al. in the battery of 10/787,749 in order to suppress the vaporization of the electrolyte.

This is a <u>provisional</u> obviousness-type double patenting rejection.

7. Claims 1-3 are provisionally rejected on the ground of nonstatutory obviousnesstype double patenting as being unpatentable over claims 1 and 4-6 of copending Application No. 10/785,970 in view of Sano et al.

Claims 1 and 4-6 of 10/785,970 teach the blended electrolyte of the instant invention but fail to teach a separator having a specific melting point.

Sano et al. teach the use of a separator in a battery cell that is capable of withstanding high temperatures ([0015]). Sano et al. teach that polyphenylene sulfide may be used as the separator, the same material used as the separator in the specification of the instant invention (claim 4 of Sano et al.).

The separator of Sano et al. is desirable since it is capable of withstand high temperatures in order to suppress the vaporization of the electrolyte ([0015]).

It would be desirable to use the polyphenylene sulfide separator of Sano et al. in the battery of 10/785,970 in order to suppress the vaporization of the electrolyte.

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Therefore, it would have been obvious to one having ordinary skill in the art at the time the invention was made to use the polyphenylene sulfide separator of Sano et al. in the battery of 10/785,970 in order to suppress the vaporization of the electrolyte.

This is a provisional obviousness-type double patenting rejection.

Response to Arguments

- 8. Applicant's arguments filed October 6, 2006 with respect to the rejection of the separator claim 1 have been fully considered and are persuasive. Therefore, the rejection has been withdrawn. However, upon further consideration, a new ground of rejection is made in view of the combination of Hamrock et al. and Sano et al. (see above).
- 9. Applicant's arguments filed October 6, 2006 concerning the rejection of the non-aqueous solvent of the electrolyte of claim 1 have been fully considered but they are not persuasive. As Applicants state on page 6 of the remarks from October 6, 2006, Hamrock et al. disclose DGM as a suitable aprotic liquid (column 13 line 52 to column 14 line 9). If DGM were used and the solvent were not blended, as is disclosed by Hamrock et al., then it would be the main component of the non-aqueous solvent as well as the only component, making it 100% by volume of the total volume of the non-aqueous solvent.

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Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Alix Elizabeth Echelmeyer whose telephone number is 571-272-1101. The examiner can normally be reached on Mon-Fri 7-4:30.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's trainer, Susy N. Tsang-Foster can be reached on 571-272-1293. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

Alix Elizabeth Echelmeyer

Examiner

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Aug Jeang Toster

aee